

1 Selective Laser Etching technology for reconfigurable microfluidic 2 and electrochemistry-on-chip

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7 **ABSTRACT**

9 Polydimethylsiloxane (PDMS) is widely used in academic microfluidics due to its favorable
10 biocompatible properties and compatibility with soft lithography. Moreover, the recent developments
11 of reconfigurable microfluidics rely on the microfabrication of sliding elements, which are 3D objects
12 insertable inside a microfluidic chip to provide a given function. However, the complexity of
13 microfluidic device geometries or sliding elements remains largely limited by the traditional
14 microfabrication methods such as, among others, on SU-8 photolithography or dry epoxy films. Such
15 methods are suited for simple, planar “2.5D” structures with uniform depths, yet struggle to produce
16 more advanced architectures due to material and alignment constraints. To address these limitations,
17 we propose to investigate advanced laser-based approaches, such as Direct Laser Writing (DLW) and
18 Selective Laser Etching (SLE), which enable the construction of high-resolution, 3D designs. We further
19 explore the use of SLE to fabricate sliding elements that enhance chip functionality, including chamber
20 reconfiguration and biological sample manipulation. As a proof of concept, we show that these
21 elements can be functionalized into pH microsensors, paving the way to reusable and reconfigurable
22 electrochemistry-on-chip.

23

24

25 **INTRODUCTION**

26 In the world of academic microfluidics, silicone-based polymers, particularly polydimethylsiloxane
27 (PDMS), have long been a go-to material for chip fabrication. Its popularity stems from a unique
28 combination of physical and chemical characteristics: optical clarity across the visible spectrum,
29 minimal autofluorescence, elasticity, permeability to gases, and biocompatibility^{1–5}. The microfluidic
30 chips fabrication process typically involves pouring liquid PDMS onto a mold, where it cures and
31 conforms with high accuracy to fine features. This microreplication protocol, designed as soft
32 lithography^{6–8}, allows for transferring the micro- to nanoscale structures from the mold⁹, thanks to the
33 flexibility of the PDMS and its minimal shrinkage. Once cured, the polymer can be bonded to glass,
34 silicon, or additional PDMS layers, often using plasma treatment, to construct sealed microchannels.

35 Traditional mold fabrication often depends on conventional or multi-level photolithography
36 techniques and commonly used materials include SU-8 patterned substrates, dry films or various
37 thermoplastics¹⁰. Such additive fabrication methods are well-suited for creating straightforward
38 designs, such as linear channels, basic geometric features and uniform channel depth, but are not
39 suitable for more complex microchannels geometries, typically due to alignment difficulties.
40 Consequently, the microchannels produced using these techniques are not really 3D and characterized
41 by features like straight-edge, rectangular profiles and uniform heights. It is possible to achieve slightly
42 more complex shapes, such as hemispherical channels or varying channel depths^{11,12}, yet these “quasi-
43 3D” geometries are typically limited and require additional steps when using conventional fabrication
44 approaches.

45 Moreover, the advent of reconfigurable microfluidic rely on the use of sliding elements to bring novel
46 functions inside a PDMS chip^{1–4}. Such elements are microfabricated rods that can be inserted inside
47 microfluidic systems, enabling a variety of applications: creation of tailored and reconfigurable culture
48 chambers, biological sample loading and retrieval, biological sample placement and aspiration,
49 creation of pressurized culture chambers. Currently developed sliding elements are usually produced
50 using epoxy resist-based materials, which have the disadvantage to be auto-fluorescent. As a result,
51 their fabrication process encounters similar issues to conventional photolithography ones, and the
52 structures generally remain quasi-3D to achieve high resolution².

53 To overcome such limitations, novel methods are being investigated for the fabrication of complex
54 microfluidic molds¹⁰. For instance, Selective Laser Etching (SLE) techniques use femtosecond laser
55 irradiations to induce local modifications in a transparent material, typically fused silica substrates
56 (amorphous SiO₂). These local modifications make the material more sensitive to chemical etching.
57 Thus, subsequent chemical treatment etches out laser modified areas to reveal the desired pattern.

58 Such components are becoming increasingly important in fields like sensors, microfluidics, medical
59 technology, data storage, and photonics^{13–15}. Their popularity stems from the silica exceptional
60 thermal, chemical, and mechanical resistance, combined with low thermal expansion and excellent
61 optical properties. These microstructures typically range in size from a few millimeters down to a few
62 micrometers. Other techniques like Direct Laser Writing (DLW), a method that exploits the nonlinear
63 absorption of ultrafast laser pulses in a photosensitive material called two-photon polymerization,
64 can also be used to construct highly intricate 3D features as small as 100 nanometers^{16–18}, but their use
65 as PDMS molds in particular remains limited.

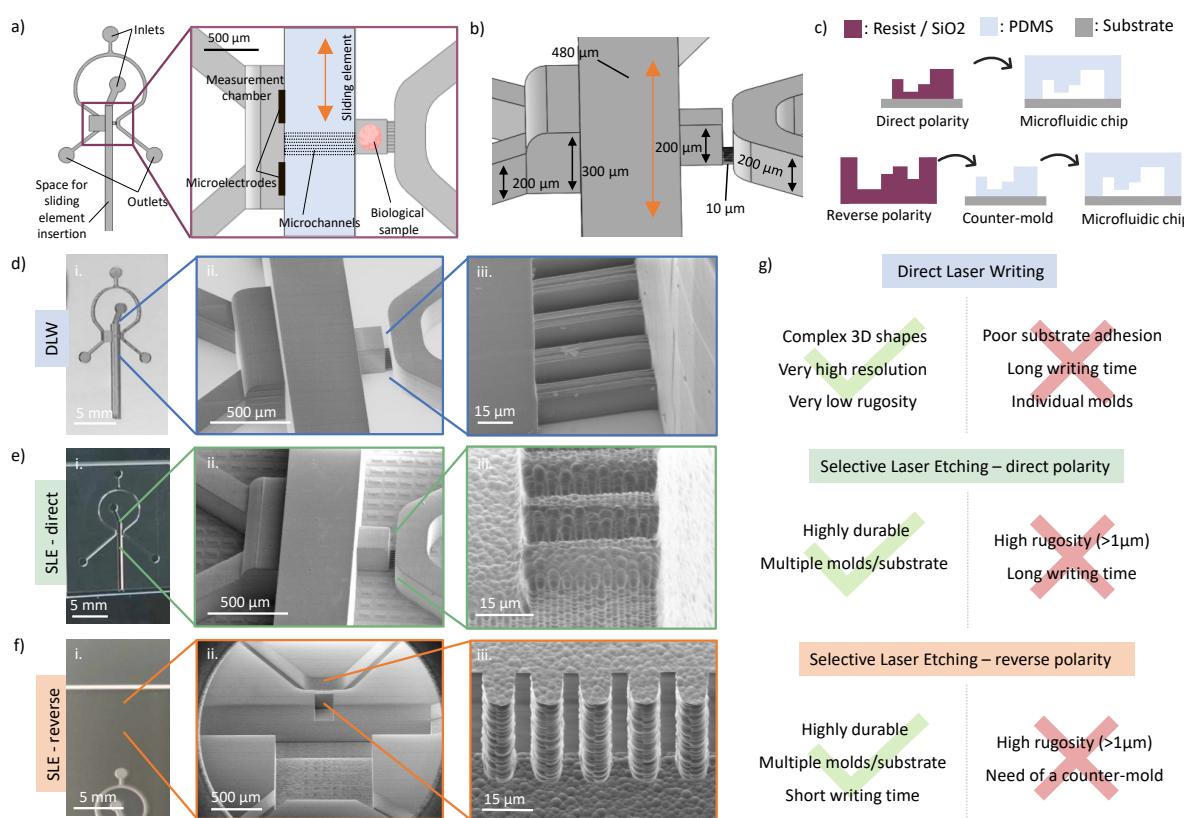
66 In this paper we propose to investigate and compare the fabrication of microfluidic molds and sliding
67 elements using both SLE and DLW methods. Such techniques have opened up exciting new possibilities
68 for designing and manufacturing microfluidic devices, ushering beyond the limitations of traditional
69 methods and enabling structures that would be extremely difficult, if not impossible, to produce with
70 conventional lithography techniques. Last, we demonstrate that basic sliding elements can be created
71 by SLE and functionalized into embedded microsensors suitable for various monitoring. As a proof of
72 concept, we integrate microelectrodes for pH measurement directly onto the sliding element, paving
73 the way for future *in situ*, reusable, real-time monitoring of biological sample metabolism¹⁹.

74 **RESULTS AND DISCUSSION**

75 ***Microfluidic system development***

76 We designed a proof-of-concept microfluidic device containing parts of highly different heights, as
77 well as round and squared edges, with a sliding element to be inserted, to investigate the possibilities
78 and limitations of DLW and SLE (Fig. 1a). This design was inspired by our recently published mechano-
79 chemostat for biological samples²⁰. Our microfluidic system featured inlets and outlets for culture
80 medium flow, a compartment for a simple or functionalized sliding element, a culture chamber for
81 introducing a biological sample via the sliding element, and a separate measurement chamber
82 designed for sensor integration, such as pH electrodes. Notably, the electrodes were positioned away
83 from the biological sample to avoid direct contact. The system was interconnected by microchannels
84 that enabled continuous medium circulation. Despite its seemingly simple design, the microfluidic
85 system incorporated channels with significant dimensional variation (see Fig1.b), ranging from 10 µm
86 × 10 µm square microchannels to a 480 µm × 480 µm compartment for the sliding element.
87 Additionally, we introduced a curved geometry to prevent the formation of "dead zones" where flow
88 and electroactive species may stagnate, potentially compromising electrochemistry measurement
89 accuracy. This curved structure mainly allowed us, at this step, to demonstrate the fabrication
90 capabilities. Conventional mold fabrication techniques were inadequate for such complex

91 architectures due to their geometric limitations and susceptibility to alignment errors, in the
 92 micrometer range, which can create leaks when using sliding elements. Consequently, we adopted
 93 advanced microfabrication approaches better suited to these design requirements.



94

95 **Figure 1: Overview of the microfluidic system design and mold fabrication developments.** a) Schematic of an
 96 example microfluidic system design, comprised of a space to insert a sliding element inside the chip, with a focus
 97 on the probing area, in the case of further integration of microelectrodes onto the sliding element. b) Schematic
 98 of the microfluidic system used in this study, highlighting the different height levels required, ranging from 10 μm
 99 to 480 μm. The curved surface of the measurement chamber can also be observed. c) Schematic of the possible
 100 polarities regarding the mold fabrication. A direct polarity allows for fabricating microfluidic chip mold in one
 101 step, whereas the reverse polarity implies the use of a counter-mold. d) Picture of the resist mold fabricated using
 102 two-photon lithography (i), with SEM images focusing on the electrodes chamber, biological sample culture
 103 chamber (ii) and microchannels (iii). e) Picture of the positive fused silica mold fabricated using selective laser
 104 etching (i), with SEM images focusing on the electrodes chamber, biological sample culture chamber (ii) and
 105 microchannels (iii). f) Picture of the negative fused silica mold fabricated using selective laser etching (i), with SEM
 106 images focusing on the electrodes chamber, biological sample culture chamber (ii) and microchannels (iii). g)
 107 Highlights of the advantages and drawbacks of each mold fabrication method.

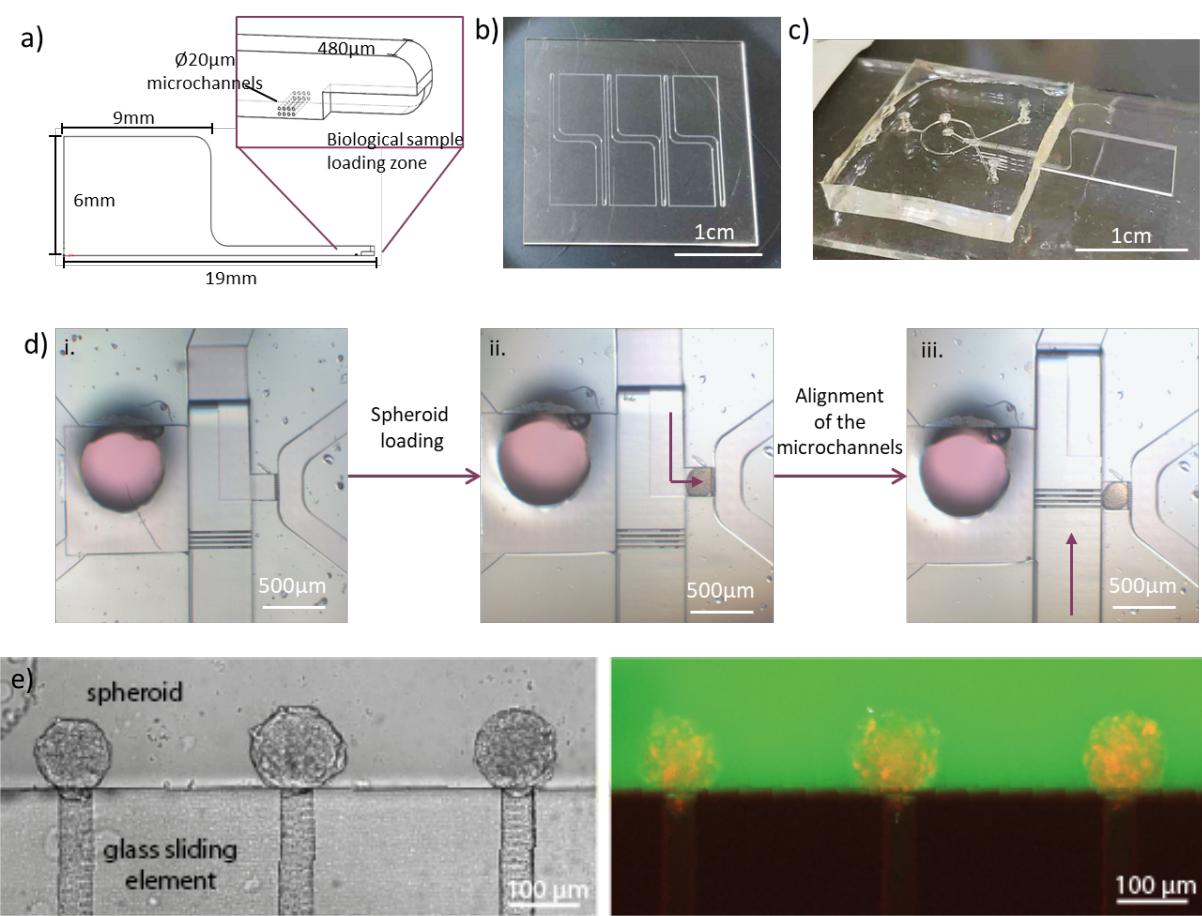
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109 Fabrication steps with DLW included depositing a 500nm layer of SU-8 resist onto the surface of a
110 silicon substrate to promote adhesion of the mold, laser-writing the direct mold structure using IP-S
111 resist (*Nanoscribe Photonic Professional GT+*), developing the piece in SU-8 developer for 10 mins and
112 strengthening the mold by 15 mins additional UV curing at 405nm to complete reticulation. The
113 obtained mold showed very low rugosity (typically below 10nm), very high resolution, and easy
114 fabrication of curved surfaces, as presented in Fig.1.d. A self-assembled monolayer of
115 perfluorodecyltrichlorosilane (FTDS) was then grafted onto the surface of the resist mold to prevent
116 PDMS adhesion. PDMS was cast onto the mold and cured at 65 °C for 4h. Once cured, the PDMS cast
117 was removed from the mold and bonded to a glass substrate (see Fig.1.c), which created the channels
118 of the microfluidic system. Yet, DLW presented some drawbacks. First, the writing time of individual
119 molds may vary depending on the objective used (10x or 25x). In this study we decided to write the
120 piece using the 25x objective, for optimal resolution. Consequently, the writing time was quite long
121 (~10 hours). In addition, the adhesion of the mold onto the substrate could be improved, as it starts
122 showing signs of detachment after a few PDMS castings (maximum 3 in our case), limiting its durability.
123 We conducted the fabrication with SLE (Nanofactory from FEMTIKA) through two different
124 approaches: direct and reverse polarity writing. The direct polarity approach involved writing the
125 negative space around the mold structure onto the silica piece, and performing subsequent chemical
126 etching to reveal the direct mold, as shown in Fig.1.e. Then, the mold was silane-treated and PDMS
127 could be casted onto the surface. Although time consuming (laser-exposing the entire negative space
128 typically took 7h, while the chemical etching step of our large piece typically takes 5h, but can be
129 parallelized), the mold obtained with this direct approach was highly durable and could be used for
130 many subsequent PDMS castings (> 10 in our case). We noted a roughness of below 1 μ m in our case
131 (Fig.1.e.iii). This roughness can be decreased through a second etching²¹. We note that this roughness
132 is not a limiting factor in our case, as it did not prevent de-molding of the PDMS, did not cause any
133 issues during the binding of the PDMS, and did not lead to any leakage.
134 Additional steps were required in order to conduct the reverse polarity approach (Fig.1.f). Specifically,
135 SLE was used to define the mold geometry within the substrate, and subsequent chemical etching
136 selectively removed the laser-modified regions to reveal the negative mold shape (as presented in
137 Fig.1.c). Then, the negative mold was silane-treated and PDMS was casted, to obtain a counter-mold.
138 The counter mold was subsequently silanized and PDMS was cast again. Finally, the PDMS cast was
139 bonded to a glass slide, creating culture chambers and channels. Advantages of this method included
140 significantly reducing both fabrication time (etching time remains similar): laser writing the entire piece
141 took only about one hour. However, the additional need of a counter-mold can be tedious, as counter

142 molds often needs to be replaced due to a loss in resolution over the uses. Depending on the needs,
143 direct polarity could thus be better suited, and not much more time-consuming.

144 All advantages and drawbacks of both DLW and our two approaches for SLE were summarized in
145 Fig.1.g. Depending on the durability, size/time of fabrication, rugosity demand, which is often not a
146 limiting factor (1 μ m rugosity does not limit the use of the microfluidic chip presented in Fig. 1) and
147 resolution, both techniques offer pros and cons. We believe that SLE of direct polarity of needed
148 microfluidic could offer more advantages than disadvantages, the difference in fabrication time being
149 important but not human-intensive. In the following, we focused on SLE as our method of choice for
150 microfabrication.

151 ***Fabrication of versatile glass sliding elements***



152

153 ***Figure 2: Fabrication of versatile glass sliding elements for microfluidics applications.*** a) Schematic of a simple
154 glass sliding element for inserting biological samples into the microfluidic device. b) Picture of the sliding elements
155 laser-written on a fused silica substrate. c) Picture of a glass sliding element inserted into a microfluidic device. d)
156 Optical microscopy images showing an example of biological sample (here, a spheroid) insertion into the
157 microfluidic device. First, the loading zone of the sliding element is aligned with the culture chamber (i). Then,

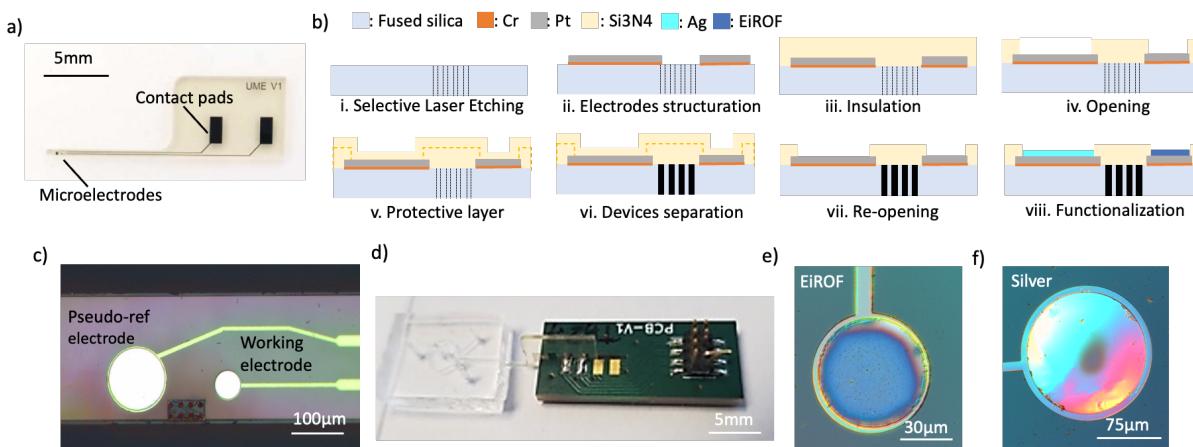
158 *flow is applied to insert the biological sample into the chamber (ii). Last, the microchannels of the sliding element*
159 *are aligned with the culture chamber, closing the chamber and allowing for culture media to flow in and out of*
160 *the chamber. e) Images of glass micropipette inserted in a PDMS chip. Spheroids are made of cells with a red-*
161 *fluorescent tag on the nucleus, and a FITC dye is added to the culture medium. We can see that the sliding element*
162 *is not autofluorescent in both the green and the red channels.*

163 We previously demonstrated the beneficial use of the sliding elements technology in microfluidic
164 systems for a variety of applications, including: creating reconfigurable confining culture chambers,
165 loading and retrieving biological samples or mimicking micro-pipettes aspiration while achieving great
166 chemical and mechanical control, and increased throughput with respect to standard aspiration²².
167 First-generation sliding elements were made out of two levels of dry film sheets²⁰. Consequently, such
168 devices could suffer from misalignment between the films, in the micrometer range, which could
169 induce leakage and / or microchannels defects. In addition, the dry film presented natural auto-
170 fluorescence due to the photoresist, which interfered with further fluorescence imaging experiments.
171 Hence, we decided to optimize the device structure and material by structuring them in fused silica
172 using SLE. Fused silica is highly durable and transparent, and that should not exhibit auto-fluorescence
173 compared with epoxy resist-based dry films. As a consequence, the optimized devices allow optical
174 inspection of the cells as well as the possibility to use optical markers in the cell culture parallel to the
175 microsensors.

176 The sliding elements fabricated here were composed of a long and thin strip with a biological sample
177 insertion and retrieval area (as illustrated on Fig.2.a). Fabrication included laser-writing the shape of
178 multiple sliding elements onto a fused silica substrate (Fig.2.b) by SLE. Such elements could then be
179 inserted into microfluidic chips, as shown in Fig.2.c, to create systems tailored for various applications.
180 For instance, spheroids could be inserted into culture chambers, as illustrated in Fig.2.d. To this aim,
181 the sliding element was inserted and the loading area was aligned with the culture chamber. Biological
182 samples immersed in culture media were then injected at the inlet and guided through the culture
183 chamber thanks to the micro flux. Last, the chamber was closed by moving the sliding element
184 upwards, and the microchannels were aligned with the culture chamber, which allowed for the media
185 to flow in the microfluidic system.

186 Additionally, we revisited the fabrication of the sliding elements used for micropipette aspiration on
187 chip using SLE²². Similarly, we showed that we could fabricate with similar solution these sliding
188 elements (Fig. 2e). Moreover, we observed no auto-fluorescence in both red and green channels
189 typically used in biology (Fig. 2e right). These results together demonstrated that SLE could be
190 efficiently used to fabricate complicated 3D shapes, from mold to sliding elements.

191 **Integration of micro-electrodes onto glass sliding elements for pH measurements.**



192

193 **Figure 3: Integration of micro-electrodes onto the sliding elements.** a) Picture of single sliding elements with
194 integrated microelectrodes. b) Schematic of the microfabrication process main steps. First, the sliding elements
195 shape are laser-written on a fused silica piece (i). Then, the electrodes are structured through a conventional
196 photolithography process, metal deposition and lift-off (ii). The surface of the device is then insulated, (iii) and
197 only the recording area of the electrodes are opened using reactive ion etching (iv). Next, a second layer of
198 insulator is deposited (v), acting as a protective layer for the metallic surfaces during the chemical etching used
199 to separate the unitary sliding elements from the fused silica substrate (vi). Finally, the working part of the
200 electrodes are re-opened via reactive ion etching (vii) and the sliding element is ready for packaging and backend
201 processes, which include electrochemical deposition to functionalize the electrodes (viii). For further details
202 regarding the microfabrication process, please refer to the materials and methods section. c) Optical microscope
203 image of the tip of the sliding element, highlighting the placement of microelectrodes around the microchannels
204 d) Picture of a sliding element welded onto a custom-made PCB. Inset: functionalized sliding element inserted into
205 a microfluidic chip. e) Optical microscope image of the working electrode functionalized with an EIROF layer. f)
206 Optical microscope image of the pseudo-reference electrode functionalized with an Ag layer.

207 We proposed to convert basic sliding elements into embedded microsensors suitable for many
208 applications, and in particular reconfigurable electrochemistry-on-chip, to perform multi-physics
209 measurements on similar biological system. We chose as a demonstrator potential cell monitoring with
210 the development of a pH sensor, which can be used to probe metabolic changes in cells. As a proof of
211 concept, we integrated microelectrodes for pH measurement directly onto the sliding element, paving
212 the way for future in situ, reusable, real-time monitoring of biological sample metabolism¹⁹.

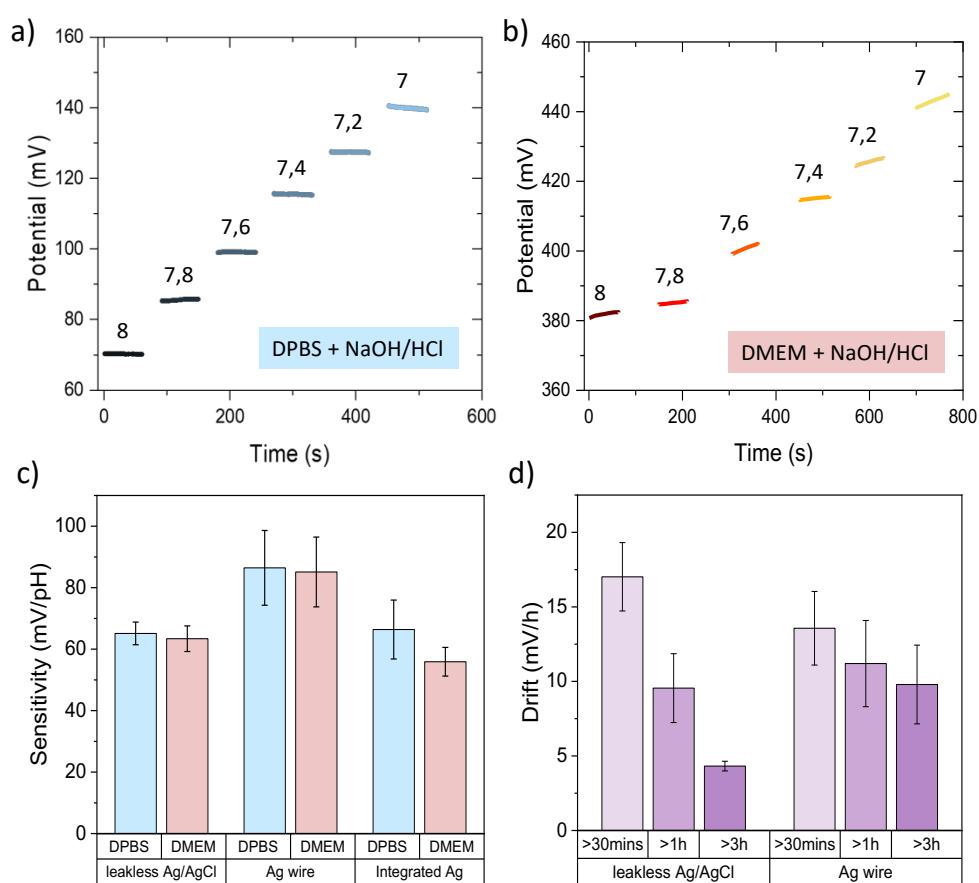
213 Fig.3.a illustrates the design of the device. The microelectrodes were located at the end of the tip, near
214 the biological sample loading area, culture chamber and connected microchannels for in situ pH
215 measurements, and the associated contact pads were placed onto the large fused-silica area. The

216 sensor comprised a working electrode (WE) ($\varnothing 100\mu\text{m}$) and a pseudo-reference electrode ($\varnothing 220\mu\text{m}$)
217 for measuring the pH in a 2-electrodes configuration, which were located above the microchannels
218 connected to the culture chamber (Fig.3.c). This fully embedded system was fabricated using a
219 combination of conventional lithography techniques and innovative approaches, as illustrated in
220 Fig.3.b. and described in the materials and methods section. First, the sliding elements shapes were
221 laser-written on a fused silica substrate. Then, the electrodes were structured through a conventional
222 photolithography process, metal deposition and lift-off. The surface of the device was then insulated
223 and only the recording area of the electrodes were opened using reactive ion etching. The elements
224 then needed to be separated, yet, the liberation of individual sliding elements required a harsh
225 chemical etching step, which would damage the exposed metallic surfaces. To overcome this
226 limitation, a second layer of $1\mu\text{m}$ of insulator was deposited, acting as a protective layer during the
227 chemical separation. Last, the working part of the electrodes was re-opened, the remaining sacrificial
228 Si₃N₄ was removed, via a second reactive ion etching process, and the unitary sliding elements were
229 integrated vertically into printed circuit boards (PCB) using micro welding, as shown in Fig.3.d. The
230 functionalized sliding elements can then be insterted inside the microfluidic chip designed in Fig. 1 (Fig.
231 3d).

232 Further backend processes are required. In fact, the WE was functionalized with an Iridium Oxide layer,
233 and the pseudo-reference electrode was covered with silver^{23,24} to perform further pH measurements.
234 Iridium oxide is a metallic material with a wide range of oxidation states²⁵, making it highly responsive
235 to its chemical environment while maintaining excellent stability in both acidic and alkaline
236 conditions²⁶. Its diverse applications include neural interfacing, pH sensing^{27,28}, catalysis²⁶, cell
237 interfacing^{29,30} or contaminant detection³¹. In this work, we focused on the electrodeposition method
238 to modify the WE with electrodeposited iridium oxide films (EIROFs) for pH measurements
239 applications. This technique offered several advantages when compared to other iridium deposition
240 processes: it was accessible, operated at room temperature, did not require costly iridium substrates,
241 and it enabled localized deposition^{32,33}. According to Baur et al.³⁴, optimal EIROF coverage could be
242 identified visually by the presence of a bright blue film. We optimized the deposition conditions (see
243 the Materials and Methods section) to meet this criterion, as shown in Fig.3.e. Regarding the
244 electrodeposition of the silver layer, we optimized the deposition conditions (see Materials and
245 Methods section), to achieve a dome-shaped, shiny and reflective layer, as illustrated in Fig.3.f.
246 Together these microfabrication steps demonstrated the first realization of an electrode on a sliding
247 element, opening the way to reconfigurable electrochemistry-on-chip.

248

249 **Performances of the electronically-active sliding elements for pH sensing.**



250

251 **Figure 4: Performances of micro-electrodes for pH sensing.** a) Graph illustrating the potential response of the
252 WE measured in an OCP configuration, using an external leakless Ag/AgCl reference electrode. The potential steps
253 correspond to different pH values in DPBS-based solutions. b) Graph illustrating the potential response of the WE
254 measured in an OCP configuration, using an external leakless Ag/AgCl reference electrode. The potential steps
255 correspond to different pH values in DMEM-based solutions. c) Bar chart showing the sensitivity of the WE, in
256 mV/pH, depending on solution and reference electrode used. d) Bar chart indicating the drift of the measured
257 potential over time in DMEM, when using external reference electrodes.

258 To characterize the sensitivity of our sensor, we measured the open circuit potential (OCP) between
259 the working electrode and a reference electrode across the range of pH values 8-6 with 0.2 pH
260 decrements. We specifically examined how the choice of reference electrode influenced the
261 measurements, as part of our effort to assess the feasibility of a fully embedded sensing system. For
262 this purpose, we compared the sensor's potential response using three different reference
263 configurations: a commercial leakless Ag/AgCl electrode (LF-1.6-48, Innovative Instruments, Inc.), a
264 bare silver wire ($\varnothing 500 \mu\text{m}$), and the integrated electroplated Ag pseudo-reference electrode.
265 Measurements were conducted in DPBS-based solutions, with pH adjusted by incremental additions

266 of 0.1M NaOH or 0.1M HCl. To evaluate performance in conditions closer to biological environments,
267 we also repeated the measurements in unbuffered DMEM-based solutions, similarly pH-adjusted,
268 anticipating future applications in complex culture media. DMEM is the basic culture medium used in
269 cell biology. We focused on alkaline-to-acid sequences, to mimic the cells production of lactate, by
270 decrementing the pH value of 0.2 every 2 minutes at 25°C. Fig.4.a and Fig.4.b illustrate the OCP using
271 the external leakless Ag/AgCl reference electrode, in DPBS- and DMEM-based solutions across the pH
272 range 8-7. In DPBS-based solutions, the measured potential remained stable at each pH value, whereas
273 in unbuffered DMEM, a slight drift was observed, likely due to more complex chemical interactions
274 and the presence of electroactive species in the medium. Fig.4.c summarizes the sensitivity obtained
275 for each condition (reference used and solution), and results appeared to be dependent on the
276 reference electrode rather than the solution. For instance, when using the leakless Ag/AgCl electrode,
277 we obtained sensitivities of $65 \pm 4\text{mV/pH}$ ($n=6$) in DPBS and $63 \pm 4\text{mV/pH}$ ($n=3$) in DMEM. With the Ag
278 wire, we obtained $86 \pm 12\text{mV/pH}$ ($n=6$) in DPBS and $85 \pm 11\text{mV/pH}$ ($n=7$) in DMEM. Last, with the
279 integrated pseudo-reference electrode we obtained $66 \pm 10\text{mV/pH}$ ($n=3$) in DPBS and $56 \pm 5\text{mV/pH}$
280 ($n=3$) in DMEM.

281 To further evaluate the performance of our electrodes, we investigated signal drift over time under
282 physiological conditions, anticipating their future use in complex culture media. Our goal was to assess
283 the measurement stability, with particular attention to the first three hours, an important window
284 during which significant pH fluctuations were expected due to the very small media volumes typically
285 used in microfluidic chips. For this purpose, the WE was immersed in complete culture medium (pH 8;
286 full composition detailed in *Materials and Methods*) for up to four hours. After an initial 30-minute
287 stabilization period, we quantified the potential drift over three intervals: from 30 minutes to 1 hour,
288 after 1 hour, and after 3 hours of immersion, as shown in Fig.4.d. The following results were obtained:
289 When using the external leakless Ag/AgCl electrode, drifts were equal to $17 \pm 2\text{mV/h}$ ($n=4$), $9 \pm 2\text{mV/h}$
290 ($n=3$) and $4.3 \pm 0.3\text{mV/h}$ ($n=4$) for 1 hour, after 1 hour, and after 3 hours of immersion, respectively.
291 The measurement drift seemed to stabilize over the first 3h to reach a value of 4mV/h . Note that the
292 maximum drift remained about 10% of the measurement over short timescale, to drop to a few
293 percents at longer timescales. On the other hand, when measuring the OCP between the WE and a
294 simple Ag wire, drifts were equal to $13 \pm 2\text{mV/h}$ ($n=5$), $11 \pm 3\text{mV/h}$ ($n=5$) and $10 \pm 3\text{mV/h}$ ($n=4$) for 1
295 hour, after 1 hour, and after 3 hours of immersion, respectively. The stabilization of the drift appears
296 steady yet much slower when compared to commercial reference electrodes.

297

298

299 **Conclusion and perspectives**

300 In this paper, we presented the use of DLW and SLE as alternatives to create both high resolution molds
301 and sliding elements for reconfigurable microfluidic devices. While DLW provided the higher resolution
302 and lower rugosity, it took longer to create molds and the mold detached after some use. Conversely,
303 SLE, in particular if performed on the reverse polarity, was quick and highly durable. Both DLW and SLE
304 allowed for the fabrication of complex 3D molds, with curved surfaces, and different heights.

305 Using SLE, we demonstrated that we could also fabricate sliding elements, to be inserted inside a
306 microfluidic device. Moreover, we demonstrated that SLE could be coupled to a classical micro-
307 electronic process and electrodeposition to create functionalized sliding elements for reconfigurable
308 electrochemistry-on-chip: the sliding elements can be written on a glass wafer first, then classical
309 planar electro-fabrication can be performed before etching the sliding elements. A great attention had
310 to be paid on the compatibility and sequence of the different fabrication steps, especially with regards
311 to the KOH used at the end of the process to detach the sliding elements from the wafer. Yet, we
312 demonstrated that this limitation can be overcome by the addition of a protective layer, to prevent
313 damaging the surfaces during this last etching step.

314 Our sensors overall exhibited a very high sensitivity compared to the literature, as other IrOx-based pH
315 sensor usually exhibit responses ranging from near-Nernstian (around 50mV/pH) to super-Nernstian
316 (above 59mV/pH and up to 69mV/pH) at 25°C^{27,35-39}. Yet, very few pH measurements studies had been
317 conducted using micro electrodes and were usually performed with electrodes of greater area⁴⁰,
318 hence, results regarding microelectrodes were quite novel. Nonetheless, studies performed on
319 microelectrodes (from Ø10µm tips to 50x100µm rectangles) also exhibited a super-Nernstian
320 sensitivity of 70-80mV/pH⁴¹ and 74.2mV/pH⁴², respectively.

321 Overall, the drift values we obtained were higher than those reported in literature, however typically
322 measured on electrodes of greater area. Nonetheless, Nguyen et al. performed studies on
323 microelectrodes (50x100µm) and obtained a drift of 72mV/h at pH 7.78 which is higher than what our
324 microsensor exhibits. Overall, studies that have explored pH measurements using Iridium-coated
325 microelectrodes, usually rely on larger-area electrodes and bulky reference systems such as calomel
326 electrodes. As a result, our findings represent a novel contribution to the field, particularly in the
327 context of miniaturized, embedded sensing platforms.

328 To conclude, we showed in the paper that SLE is a very good candidate to create novel 3D-molds and
329 sliding elements and that these ones can be functionalized to create electrodes. The combination of

330 these two leads to reconfigurable microfluidic and electrochemistry-on-chip: while the sliding element
331 can bring novel functions that can be coupled to deformable elements inside a microfluidic device to
332 mechanically and chemically control the biological sample, the inserted electrode can be used to
333 perform additional measurements, paving the way to multi-physics analyses of biological samples.

334 **MATERIALS AND METHODS**

335 ***Direct laser writing with Nanoscribe***

336 Our direct laser writing process operates on two-photon exposure. To manage this technique, we use
337 Photonic Professional GT+ equipment and IP-S resin (Nanoscribe). The main process parameters are
338 fixed with Describe software (Nanoscribe) and an STL file of the mold. The total height of the mold is
339 480 μm . It is sliced with 240 layers. The mold's length and width are 13.6 mm and 6.5 mm, respectively.
340 It is assembled from 697 blocks. Each block is scanned by a pulsed infrared laser with a 25x optical lens.
341 The block connection positions and block overlaps are optimized to increase the mold quality. The
342 main options for configuring block writing are:

343 - Hatch spacing: 0.5 μm
344 - Writing mode: shell and scaffold
345 - Interface detection: automatic for critical blocks with small patterns
346 - Writing speed: 100 mm/s

347 ***Selective laser etching with Femtika Laser Nanofactory***

348 We used a Femtika Laser Nanofactory system (Femtika.Ltd, Lithuania), equipped with an ultrafast
349 Yb:KGW femtosecond laser (1030 nm central wavelength, 700 fs pulse duration, 610 kHz repetition
350 rate), to expose glass substrates.

351 High-precision XYZ translation stage combined with galvanometric scanners enables to focus the laser
352 beam into the volume of the glass substrate at high precision. We used a 20x microscope objective
353 with a numerical aperture (NA) of 0.45, which produced a point spread function of about 3 μm in
354 diameter in the XY plane and 14 μm along the Z-axis. The writing process was performed in an ambient
355 air condition, on a vibration-isolated stage to maintain alignment stability.

356 After the laser-induced modification process, the substrate was plunged into a potassium hydroxide
357 (KOH) solution (6M) at 90 °C for 5 hours. Following this step, the substrate cooled down in the oven
358 until it reached 50 °C. The substrate was cleaned with deionized (DI) water and left to dry in ambient
359 conditions.

360 ***Microfabrication process of the electronically-active sliding element***

361 First, the sliding elements shape and channels are laser-written on a fused silica piece (Nanofactory
362 Femtika), as illustrated in Fig.2.b. The fused silica substrate is subsequently cleaned using buffered
363 oxide etch (BOE 7:1, *MicroChemicals GmbH*), to remove any silica dust or particles resulting of a small
364 ablation during the writing when the laser beam is at the interface air/glass. Then, the electrodes
365 shape, contact pads and access lines are structured using photolithography (*MABA6Gen4*, *Suss*
366 *Microtec*), subsequent Cr (50nm) and Pt (200nm) physical vapor deposition (*Eva600*, *Alliance Concept*)
367 and lift-off of the resist in acetone. Then an insulation layer of Si_3N_4 (1 μm) is deposited using plasma-
368 enhanced chemical vapor deposition (*PECVD100*, *ApSy*). thickness of the insulation layer allows for
369 both electronical insulation for the liquid media and mechanical shield as the sliding-element will
370 further be inserted into and retrieved from the microfluidic system. Next, the recording area of the
371 electrodes are opened using reactive ion etching (*SI500-DRIE*, *Sentech*). The next step consists in
372 separating the sliding elements from the fused silica substrates using a KOH-based chemical etchant.
373 Prior to this step the sliding elements are covered in an additional 1 μm of Si_3N_4 , acting as a sacrificial
374 layer, to protect the metallic surfaces from the etchant. Last, the residual Si_3N_4 is removed from the
375 recording area of both electrodes through the same plasma etching process, and unitary sliding
376 elements are ready for packaging. To this aim, each device is inserted in the slit of our custom-designed
377 PCB and contact pads are welded, as shown on Fig.3.d.

378 Focused-ion beam (FIB) milling and scanning electron microscopy (SEM) were performed with a
379 HeliosNanoLab 600i Dual Beam (*FEI*) for characterization of the surfaces.

380 ***Functionalization of the working electrode***

381 To perform EIROF deposition we used the solution developed by Marzouk et al.⁴³, following the
382 pioneering work of Yamanaka et al.⁴⁴. The EIROF depositions are performed in a conventional 3-
383 electrode setup, with a Pt wire as the counter electrode, an external commercial leakless Ag/AgCl
384 electrode as the reference electrode (*LF-1.6-48*, *Innovative Instruments, Inc.*) and the integrated Pt
385 working electrode. We use the potentiostatic route, with an applied potential of 0.6V until the desired
386 charge of 25mC/cm² is consumed.

387 ***Functionalization of the pseudo reference electrode***

388 To perform silver deposition, we used a commercial electrolyte solution (*Silver electrolyte, TIFOO.*) and
389 silver has been deposited using the galvanostatic route in a 3-electrode configuration as described
390 above. We applied a current of 1 μA for 15mins.

391

392 ***Solutions***

393 pH measurements were conducted either in DPBS-based (ThermoFisher) solutions or in unbuffered
394 DMEM-based solutions (ThermoFisher), and were pH adjusted by adding small amounts of 0.1M NaOH,
395 or 0.1M HCl. The complete culture media used for drift measuring is composed of unbuffered DMEM
396 supplement with fetal bovine serum (FBS), penicillin / streptomycin, glutamax and pyruvate.

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406 **BIBLIOGRAPHY**

407 1. Firpo, G., Angeli, E., Repetto, L. & Valbusa, U. Permeability thickness dependence of
408 polydimethylsiloxane (PDMS) membranes. *Journal of Membrane Science* **481**, 1–8 (2015).

409 2. Charati, S. G. & Stern, S. A. Diffusion of Gases in Silicone Polymers: Molecular Dynamics
410 Simulations. *Macromolecules* **31**, 5529–5535 (1998).

411 3. Miranda, I. *et al.* Properties and Applications of PDMS for Biomedical Engineering: A Review.
412 *Journal of Functional Biomaterials* **13**, 2 (2022).

413 4. Schneider, F., Draheim, J., Kammerer, R. & Wallrabe, U. Process and material properties of
414 polydimethylsiloxane (PDMS) for Optical MEMS. *Sensors and Actuators A: Physical* **151**, 95–99
415 (2009).

416 5. Sales, F. C. P., Ariati, R. M., Noronha, V. T. & Ribeiro, J. E. Mechanical Characterization of PDMS
417 with Different Mixing Ratios. *Procedia Structural Integrity* **37**, 383–388 (2022).

418 6. Morbioli, G. G., Speller, N. C. & Stockton, A. M. A practical guide to rapid-prototyping of PDMS-
419 based microfluidic devices: A tutorial. *Analytica Chimica Acta* **1135**, 150–174 (2020).

420 7. Fujii, T. PDMS-based microfluidic devices for biomedical applications. *Microelectronic Engineering*
421 **61–62**, 907–914 (2002).

422 8. Zhao, X.-M., Xia, Y. & Whitesides, G. M. Soft lithographic methods for nano-fabrication. *Journal of*
423 *Materials Chemistry* **7**, 1069–1074 (1997).

424 9. Submicrometer resolution replication of relief patterns for integrated optics | *Journal of Applied*
425 *Physics* | AIP Publishing. <https://pubs.aip.org/aip/jap/article-abstract/45/10/4557/507323/Submicrometer-resolution-replication-of-relief>.

427 10. Venzac, B. Light-based 3D printing and post-treatments of moulds for PDMS soft lithography.
428 *Lab on a Chip* <https://doi.org/10.1039/D4LC00836G> (2025) doi:10.1039/D4LC00836G.

429 11. Freitas, D. N., Mongersun, A., Chau, H. & Araci, I. E. Tunable soft lithography molds enable
430 rapid-prototyping of multi-height channels for microfluidic large-scale integration. *J. Micromech.*
431 *Microeng.* **29**, 035009 (2019).

432 12. Unger, M. A., Chou, H.-P., Thorsen, T., Scherer, A. & Quake, S. R. Monolithic Microfabricated
433 Valves and Pumps by Multilayer Soft Lithography. *Science* **288**, 113–116 (2000).

434 13. Gottmann, J., Hermans, M., Repiev, N. & Ortmann, J. Selective Laser-Induced Etching of 3D
435 Precision Quartz Glass Components for Microfluidic Applications—Up-Scaling of Complexity and
436 Speed. *Micromachines* **8**, 110 (2017).

437 14. Kluck, S. *et al.* Replicative manufacturing of metal moulds for low surface roughness polymer
438 replication. *Nat Commun* **13**, 5048 (2022).

439 15. Butkutė, A. *et al.* Femtosecond Laser Assisted 3D Etching Using Inorganic-Organic Etchant.
440 *Materials (Basel)* **15**, 2817 (2022).

441 16. Su, R., Wang, F. & McAlpine, M. C. 3D printed microfluidics: advances in strategies,
442 integration, and applications. *Lab Chip* **23**, 1279–1299 (2023).

443 17. M. Young, O., Xu, X., Sarker, S. & D. Sochol, R. Direct laser writing-enabled 3D printing
444 strategies for microfluidic applications. *Lab on a Chip* **24**, 2371–2396 (2024).

445 18. Li, Q. *et al.* Direct 3D-printing of microlens on single mode polarization-stable VCSEL chip for
446 miniaturized optical spectroscopy. *JOM* **3**, 033501 (2023).

447 19. Weltin, A., Enderle, B., Kieninger, J. & Urban, G. A. A novel, multiparametric, flexible
448 microsensor for metabolic monitoring in vivo. in *2013 IEEE SENSORS 1–3* (IEEE, Baltimore, MD,
449 USA, 2013). doi:10.1109/ICSENS.2013.6688341.

450 20. Ben Meriem, Z. *et al.* A microfluidic mechano-chemostat for tissues and organisms reveals
451 that confined growth is accompanied with increased macromolecular crowding. *Lab Chip* **23**,
452 4445–4455 (2023).

453 21. Widmer, R. N. *et al.* Smooth or not: Robust fused silica micro-components by femtosecond-
454 laser-assisted etching. *Materials & Design* **204**, 109670 (2021).

455 22. Landiech, S. *et al.* Parallel on-chip micropipettes enabling quantitative multiplexed
456 characterization of vesicle mechanics and cell aggregates rheology. *APL Bioeng* **8**, 026122 (2024).

457 23. Korostynska, O. Review on State-of-the-art in Polymer Based pH Sensors. (2007).

458 24. Popovtzer, R., Neufeld, T., Ron, E. Z., Rishpon, J. & Shacham-Diamand, Y. Electrochemical
459 detection of biological reactions using a novel nano-bio-chip array. *Sensors and Actuators B: Chemical* **119**, 664–672 (2006).

460 25. Pyykkö, P. & Xu, W.-H. The Formal Oxidation States of Iridium Now Run from –III to +IX. *Angew. Chem. Int. Ed.* **54**, 1080–1081 (2015).

461 26. Quinson, J. Iridium and IrO_x nanoparticles: an overview and review of syntheses and
462 applications. *Advances in Colloid and Interface Science* **303**, 102643 (2022).

463 27. Huang, W.-D., Cao, H., Deb, S., Chiao, M. & Chiao, J. C. A flexible pH sensor based on the
464 iridium oxide sensing film. *Sensors and Actuators A: Physical* **169**, 1–11 (2011).

465 28. Li, J., Du, Y. & Fang, C. Developing an Iridium Oxide Film Modified Microelectrode for
466 Microscale Measurement of pH. *Electroanalysis* **19**, 608–611 (2007).

467 29. Kim, Y. H., Koo, H., Kim, M. S. & Jung, S.-D. Iridium oxide on indium-tin oxide nanowires: An
468 all metal oxide heterostructured multi-electrode array for neuronal interfacing. *Sensors and
469 Actuators B: Chemical* **273**, 718–725 (2018).

470 30. Negi, S., Bhandari, R., Rieth, L. & Solzbacher, F. *In vitro* comparison of sputtered iridium oxide
471 and platinum-coated neural implantable microelectrode arrays. *Biomed. Mater.* **5**, 015007 (2010).

472 31. Wang, J., Yokokawa, M., Satake, T. & Suzuki, H. A micro IrO_x potentiometric sensor for direct
473 determination of organophosphate pesticides. *Sensors and Actuators B: Chemical* **220**, 859–863
474 (2015).

475 32. Meyer, R. D., Cogan, S. F., Nguyen, T. H. & Rauh, R. D. Electrodeposited iridium oxide for
476 neural stimulation and recording electrodes. *IEEE Trans. Neural Syst. Rehabil. Eng.* **9**, 2–11 (2001).

477 33. Eddarir, A. Conception et réalisation d'un dispositif tridimensionnel à matrice de
478 microélectrodes pour la mesure électrophysiologique et la détection pH d'une culture neuronale
479 in vitro. (Toulouse 3, 2022).

480 34. Baur, J. E. & Spaine, T. W. Electrochemical deposition of iridium (IV) oxide from alkaline
481 solutions of iridium(III) oxide. *Journal of Electroanalytical Chemistry* **443**, 208–216 (1998).

484 35. Grant, S. A., Bettencourt, K., Krulevitch, P., Hamilton, J. & Glass, R. In vitro and in vivo
485 measurements of fiber optic and electrochemical sensors to monitor brain tissue pH. *Sensors and*
486 *Actuators B: Chemical* **72**, 174–179 (2001).

487 36. Hung Cao *et al.* An Implantable, Batteryless, and Wireless Capsule With Integrated
488 Impedance and pH Sensors for Gastroesophageal Reflux Monitoring. *IEEE Trans. Biomed. Eng.* **59**,
489 3131–3139 (2012).

490 37. Ghoneim, M. T. *et al.* Recent Progress in Electrochemical pH-Sensing Materials and
491 Configurations for Biomedical Applications. *Chem. Rev.* **119**, 5248–5297 (2019).

492 38. Bene, C. Développement de micro(bio)capteurs électrochimiques en vue de leur intégration
493 dans une plateforme microfluidique (Lab-On-Disc) pour le contrôle in situ de la qualité des eaux
494 douces de surface. (Toulouse 3, 2023).

495 39. Ratanaporncharoen, C. *et al.* pH Mapping on Tooth Surfaces for Quantitative Caries Diagnosis
496 Using Micro Ir/IrO_x pH Sensor. *Anal Chem* **90**, 4925–4931 (2018).

497 40. Chawang, K., Bing, S. & Chiao, J.-C. Printable and Flexible Iridium Oxide-Based pH Sensor by a
498 Roll-to-Roll Process. *Chemosensors* **11**, 267 (2023).

499 41. Iridium oxide pH microelectrode - VanHoudt - 1992 - Biotechnology and Bioengineering -
500 Wiley Online Library.
501 <https://analyticalsciencejournals.onlinelibrary.wiley.com/doi/10.1002/bit.260400507>.

502 42. Nguyen, C. M. *et al.* Micro pH Sensors Based on Iridium Oxide Nanotubes. *IEEE Trans.*
503 *Nanotechnology* **13**, 945–953 (2014).

504 43. Marzouk, S. A. M. Improved Electrodeposited Iridium Oxide pH Sensor Fabricated on Etched
505 Titanium Substrates. *Anal. Chem.* **75**, 1258–1266 (2003).

506 44. Yamanaka, K. Anodically Electrodeposited Iridium Oxide Films (AEIROF) from Alkaline
507 Solutions for Electrochromic Display Devices. *Jpn. J. Appl. Phys.* **28**, 632–637 (1989).

508